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(21) International Application Number: PCT/US96/04976 (22) International Filing Date: 10 April 1996 (10.04.96) (30) Priority Data: 08/419,814 11 April 1995 (11.04.95) US (71) Applicant: GREAT LAKES CHEMICAL CORPORATION [US/US]; Highway 52 N.W., P.O. Box 2200, West Lafayette, IN 47906 (US). (72) Inventors: ROSE, Richard, S.; 5841 Lookout Drive, West Lafayette, IN 47906 (US). BHATTACHARYA, Bhabatosh; 1909 Arcadia Drive, Lafayette, IN 47905 (US). FAVSTRIT-SKY, Nicolai, A.; 444 Southern Drive, Lafayette, IN 47905 (US). (74) Agents: HENRY, Thomas, Q. et al.; Woodard, Emhardt, Naughton, Moriarty & McNett, Bank One Center/Tower, Suite 3700, 111 Monument Circle, Indianapolis, IN 46204 (US).		(81) Designated States: BR, CA, European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE). Published <i>With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
(54) Title: BENZOATES AS FLAME RETARDANTS AND/OR PLASTICIZERS		
(57) Abstract A method of making flame retarded polyurethanes by including a bromobenzoate compound in the polyurethane mixture. Also disclosed is a method of making flame retarded polyvinyl chlorides, unsaturated polyesters and epoxy resins by incorporating a bromobenzoate compound into the polymer system. The bromobenzoate compounds of the invention may also be used as plasticizers for PVC and rubber.		

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"BENZOATES AS FLAME RETARDANTS AND/OR PLASTICIZERS"

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FIELD OF THE INVENTION

The present invention relates generally to novel uses for bromobenzoate compounds, and more particularly to their use as combustion modifiers and/or plasticizers in polymeric systems.

BACKGROUND OF THE INVENTION

Compounds containing aromatically bound bromine have previously been used as additives for polyurethanes, polyvinyl chloride, and other polymer systems such as unsaturated polyesters, epoxies, and rubbers, particularly where resistance to thermal and hydrolytic decomposition is desired. In recent years however, unsubstantiated claims that certain of these compounds (such as brominated diphenyl oxides) are hazardous to human health have resulted in a search for members of that class that are demonstrably safe and effective.

Among the requirements for such compounds are that they be flowable liquids with minimal effect upon the physical properties of the polymer. Where there is an effect on the polymer, it is desired that the effect improve the physical properties, such as plasticity.

A need therefore exists for improved combustion modifiers and/or plasticizers for polymers such as polyurethane and polyvinyl chloride. The present invention addresses that need.

SUMMARY OF THE INVENTION

Briefly describing one preferred aspect of the present invention, there is provided a method of making flame retarded polyurethane by including a bromobenzoate compound in the polyurethane mixture. In a second preferred embodiment of the present invention there is provided a method of making flame retarded polyvinyl chlorides by incorporating a bromobenzoate into the polyvinyl chloride. Flame retarded polyurethane and polyvinyl chloride compositions incorporating a flame retardant amount of a bromobenzoate compound also constitute preferred embodiments of the invention. In such methods and compositions, the bromobenzoate compounds provide surprising and unexpected compatibility with the resin systems, affording overall compositions exhibiting excellent physical and chemical properties.

One object of the present invention is to provide methods of improving the physical properties of polyurethane.

Another object of the present invention is to provide methods of improving the physical properties of polyvinyl chlorides.

Additional objects of the invention include the provision of polyurethane, polyvinyl chloride and other polymer resin compositions with surprisingly improved physical properties.

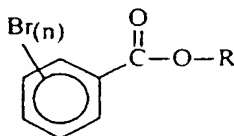
Still further objects and advantages of the invention will be apparent from the following description.

DESCRIPTION OF THE PREFERRED EMBODIMENT

For the purpose of promoting an understanding of the principles of the invention, reference will now be made to preferred embodiments and specific language will be used to describe the same. It will nevertheless be understood that no limitation of the scope of the invention is thereby intended, such alterations and further modifications in the preferred embodiments, and such further applications of the principles of the invention as illustrated herein being contemplated as would normally occur to one skilled in the art to which the invention pertains.

As previously indicated, preferred aspects of the present invention relate to methods for improving the physical characteristics of polyurethanes and polyvinyl chlorides. In particular, the present invention uses ring-brominated benzoates (*i.e.*, bromobenzoate compounds, such as tetrabromobenzoate compounds) as flame retardants and/or plasticizers in polyurethanes and polyvinyl chlorides. Generally, the bromobenzoate compounds are incorporated directly into the polymer to provide the aforementioned beneficial properties.

In this regard, preferred bromobenzoate compounds for use in the present invention include those encompassed by formula (I):



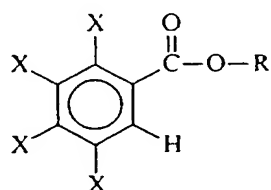
(I)

wherein n is an integer from 1 to 5, and R is an organic group having up to about 30 carbon atoms, and optionally being substituted with one or more groups such as alkoxy, halo (especially bromo or chloro), thio, and the like. In more preferred bromobenzoates, n is an integer from 1 to 4, most preferably from 2 to 4.

When the bromobenzoates of the present invention are incorporated into polyurethanes, the compositions meet both the physical property and combustibility test criteria for polyurethanes without the health concerns associated with brominated diphenyl oxides. In addition, testing in PVC

applications shows bromobenzoates to have surprisingly better compatibility than the prior art compound di(2-ethylhexyl)tetrabromophthalate.

Desired bromobenzoates are preferably prepared by reacting bromophthalic anhydride with the appropriate alcohol in the presence of a decarboxylation catalyst. In one preferred embodiment the reaction is performed in an inert solvent to minimize the creation of phthalates. Benzoic acid or phthalic anhydride having different numbers of bromines on the ring, or mixtures having various levels of bromination, can also be used as starting materials in the synthesis of the benzoates. Thus, in these preparative processes, preferred bromobenzoates will be formed which are encompassed by the formula (II):



(II)

wherein each X, which may be the same as or may differ from on another, is -Br or -H, with the proviso that at least one X is -Br, and R is as defined above in formula (I). Preferred compounds of formula (II) occur where each X is Br.

Examples of bromobenzoates which can be used in the present invention are those prepared from alcohols having a boiling point of between about 160°C and about 230°C, preferably between 180°C and 205°C. These alcohols provide the advantage of allowing the decarboxylation reaction to proceed at a reasonable rate, while still allowing for ease of stripping the excess alcohol from the product at the end of the reaction. Preferably the alcohol is a non-halogenated, non-sulfur-containing, non-nitrogen-containing alcohol (and thus R will preferably be free from halogen, sulfur or nitrogen). Branched chain alcohols are most preferred.

Appropriate alcohols with boiling points between about 160°C and about 230°C include 2-(2-methoxy)ethoxyethanol, 2-butoxyethanol, 3,3-diethoxy-1-propanol, di(propylene glycol) methyl ether, 2-ethyl-1-hexanol, 3-ethyl-1-hexanol,

3,4-dimethyl-1-hexanol, 3,5-dimethyl-1-hexanol, 3-methyl-1-heptanol, 4-methyl-1-heptanol, 2-octanol, 5-methyl-1-heptanol, 3-octanol, 2-propyl-1-pentanol, 2,6-dimethyl-4-heptanol, 2-nonanol, dihydromyrcenol, 3,5,5-trimethyl-1-hexanol, 3,7-dimethyl-1-octanol, 3,7-dimethyl-3-octanol, mixed C₇-C₉ alcohols, isooctyl alcohol, mixed C₉ alcohols, 3-furanmethanol, furfuryl alcohol, tetrahydrofurfuryl alcohol, 3-acetyl-1-propanol, 2-isopropoxyethanol, 3-methoxy-1-butanol, 2-cyclohexen-1-ol, 1,5-hexadien-3-ol, t,t-2,4-hexadien-1-ol, 3-hexyn-1-ol, 5-hexyn-1-ol, cyclopentanemethanol, 4-methyl-1-pentanol, 3-(trimethylsilyl)allyl alcohol, benzyl alcohol, 3-trimethylsilyl-1-propanol, 3-cyclohexen-1-methanol, 3-methyl-2-cyclohexen-1-ol, cycloheptanol, cyclohexylmethanol, 1-methylcyclohexanol, 2-methylcyclohexanol, 3-methylcyclohexanol, 4-methylcyclohexanol, 1-heptanol, 2-heptanol, propylene glycol butyl ether, 2-methylbenzyl alcohol, 3-methylbenzyl alcohol, 4-methylbenzyl alcohol, phenethyl alcohol, sec-phenethyl alcohol, 1-octyn-3-ol, cycloheptanemethanol, 2-cyclohexylethanol, 1-cyclohexylethanol, cyclooctanol, 3-cyclopentyl-1-propanol, 2,3-dimethylcyclohexanol, 2,6-dimethylcyclohexanol, 3,5-dimethylcyclohexanol, 2-ethylcyclohexanol, 4-ethylcyclohexanol, 1-octanol, 6-methyl-5-hepten-2-ol, 1-octen-3-ol, 6-methyl-2-heptanol, 2-(cyclohexyloxy)ethanol, 2,2-dimethoxycyclohexanol, 2,4,4-trimethyl-1-pentanol, 1-phenyl-1-propanol, 1-phenyl-2-propanol, 2-phenyl-2-propanol, 3-nonyl-1-ol, 2,4-dimethyl-2,6-heptadien-1-ol, 3-cyclohexyl-1-propanol, 3,5,5-trimethyl-2-cyclohexen-1-ol, 3-nonen-1-ol, 3-ethyl-2,2-dimethyl-3-pentanol, 1-nonanol, 1-myrtanol, 2-phenyl-3-butyne-2-ol, 1-phenyl-1-cyclopropanemethanol, 2-methyl-1-phenyl-2-propanol, isopulegol, linalool, 1-myrtanol, nerol, terpineol, terpinen-4-ol, citronellol, 4-cyclohexyl-1-butanol, 2-decanol, 4-decanol, mixed C₇-C₁₁ alcohols, isodecyl alcohol, hexyl decyl alcohol, 1,3-dibromo-2-propanol, 2,3-dibromopropanol, 1,3-dichloro-2-propanol, 1,3-difluoro-2-propanol, 3-bromo-1-propanol, 3-chloro-1-propanol, 4-chloro-1-butanol, 2-(methylthio)ethanol, 3-bromo-3-buten-1-ol, 3-pyrrolidinol, 1,4-dibromo-2-butanol, 2-(2-chloroethoxy)ethanol, 3-methylthio-1-propanol, 3-thiophenemethanol, 2,2-bis(chloromethyl)-1-propanol, tetrahydro-4H-pyran-4-ol, 3-bromo-2,2-dimethyl-1-propanol,

2-(3-thienyl)ethanol, 3-chloro-2,2-dimethyl-1-propanol, 1-methyl-3-pyrrolidinol, 4-(methylthio)-1-butanol, 2-(trimethylsilyl)ethanol, 2-(2-thienyl)ethanol, tetrahydropyran-2-methanol, 6-bromo-1-hexanol, 6-chloro-1-hexanol, 7-bromo-1-heptanol, N,N-diethylethanolamine, 1-methyl-2-pyrrolidinemethanol, 1-piperideneethanol, 3-(methylthio)-1-hexanol, 3-diethylamino-1-propanol, 2-(diisopropylamino)ethanol and 2- ϵ -(2-(dimethylamino)ethyl)methylaminoethanol. Of these, 3-furanmethanol, furfuryl alcohol, tetrahydrofurfuryl alcohol, 3-acetyl-1-propanol, 2-isopropoxyethanol, 3-methoxy-1-butanol, 2-cyclohexen-1-ol, 1,5-hexadien-3-ol, t,t-2,4-hexadien-1-ol, 3-hexyn-1-ol, 5-hexyn-1-ol, cyclopentanemethanol, 4-methyl-1-pentanol, 3-(trimethylsilyl)allyl alcohol, cyclohexylmethanol, 3-trimethylsilyl-1-propanol, benzyl alcohol, 3-cyclohexen-1-methanol, 3-methyl-2-cyclohexen-1-ol, cycloheptanol, 1-methylcyclohexanol, 2-methylcyclohexanol, 3-methylcyclohexanol, 4-methylcyclohexanol, 1-heptanol, 2-heptanol, propylene glycol butyl ether, 2-methylbenzyl alcohol, 3-methylbenzyl alcohol, 4-methylbenzyl alcohol, phenethyl alcohol, sec-phenethyl alcohol, 1-octyn-3-ol, cycloheptanemethanol, 2-cyclohexylethanol, 1-cyclohexylethanol, cyclooctanol, 3-cyclopentyl-1-propanol, 2,3-dimethylcyclohexanol, 2,6-dimethylcyclohexanol, 3,5-dimethylcyclohexanol, 2-ethylcyclohexanol, 4-ethylcyclohexanol, 6-methyl-5-hepten-2-ol, 1-octen-3-ol, 2-(cyclohexyloxy)ethanol, 2,2-dimethoxycyclohexanol, 6-methyl-2-heptanol, 1-octanol, 2,4,4-trimethyl-1-pentanol, 1-phenyl-1-propanol, 1-phenyl-2-propanol, 2-phenyl-2-propanol, 2,4-dimethyl-2,6-heptadien-1-ol, 3-nonyl-1-ol, 3,5,5-trimethyl-2-cyclohexen-1-ol, 3-cyclohexyl-1-propanol, 3-nonen-1-ol, 1-nonanol, 3-ethyl-2,2-dimethyl-3-pentanol, 2-phenyl-3-butyne-2-ol, 1-phenyl-1-cyclopropanemethanol, 1-myrtanol, isopulegol, 2-methyl-1-phenyl-2-propanol, linalool, 1-myrtanol, nerol, terpineol, terpinen-4-ol, citronellol, 2-decanol, 4-decanol, 4-cyclohexyl-1-butanol, mixed C₇-C₁₁ alcohols, isodecyl alcohol and hexyl decyl alcohol are preferred, while 3,3-diethoxy-1-propanol, 2-(2-methoxy)ethoxyethanol, 3,4-dimethyl-1-hexanol, 2-butoxyethanol, di(propylene glycol) methyl ether, 3-ethyl-1-hexanol, 2-ethyl-1-hexanol, 3,5-dimethyl-1-hexanol, 3-methyl-1-heptanol, 4-methyl-1-heptanol, 5-methyl-1-

heptanol, 2-octanol, 3-octanol, 2-propyl-1-pentanol, 2,6-dimethyl-4-heptanol, 2-nonanol, 3,5,5-trimethyl-1-hexanol, dihydromyrcenol, 3,7-dimethyl-1-octanol, 3,7-dimethyl-3-octanol, mixed C₇-C₉ alcohols, isooctyl alcohol and mixed C₉ alcohols are most preferred.

The total organic bromine content of the bromobenzoate is an important factor for efficiency as a flame retardant, and can be adjusted by the alcohol and/or solvent used in the synthesis and/or the catalyst loading and/or the number of bromines on the aromatic ring. In one aspect of the invention the total organic, aromatically-bound bromine content of the bromobenzoate falls within the range of 50-65%. In another aspect of the invention where the plasticizing nature of the bromobenzoate is of primary importance, the total organic, aromatically-bound bromine content of the bromobenzoate is between 10% and 65%.

As will be appreciated by persons skilled in the art, the bromobenzoate compound is incorporated into thermoset polymers such as polyurethanes by including the bromobenzoate in the polyurethane mixture as the polymer is prepared. This process has been referred to as the "one-shot" technique, and is described with more particularity in common reference materials such as the Modern Plastics Encyclopedia, Vol. 71, No. 12 (1994), and was used in Examples 11 through 15 below.

The incorporation of bromobenzoates into polyvinyl chlorides may be accomplished either by including the desired tetrabromobenzoate in the mixture as the polyvinyl chloride is being formed, or by incorporating the bromobenzoate into polymerized polyvinyl chloride. Specific techniques for incorporating additives such as bromobenzoates into thermoplastics such as PVC are known to the art and may be used to accomplish that step of the present invention.

As will be understood, the level of bromobenzoate incorporated into the polymer resin to provide an effective flame retarding amount will vary widely in accordance with many factors such as the particular resin used, the application contemplated, other additives present, etc. Typically, the bromobenzoate will be incorporated at levels between about 2% and 50% of the total system weight, and more commonly at levels between about 5% and 30% of the total system weight.

It will be understood that other conventional additives may also be incorporated into the polymer systems. For example, the bromobenzoate product can be incorporated along with other brominated flame retardant compounds; however, it is preferred in this regard that the bromobenzoate compound constitute a predominant portion (i.e. greater than 50% by weight) of the total amount of brominated flame retardant included in the system. Flame retardant materials such as oxides of Group V elements, especially antimony oxides, and/or phosphorous-containing compounds, can also be included. Additional conventional additives may include antioxidants, antistatic agents, colorants, fibrous reinforcements, fillers, foaming/blowing agents, catalysts, heat stabilizers, impact modifiers, lubricants, plasticizers, processing aids, UV light stabilizers, crosslinking/curing agents, etc.

In another aspect of the invention bromobenzoates such as tetrabromobenzoates are used as plasticizers for materials such as PVC and rubber. These benzoates are useful as secondary plasticizers to replace a portion of the primary plasticizer, often a phthalate, as well as plasticizers on their own, without regard to their performance as flame retardants. Thus, these bromobenzoates can be incorporated as plasticizers in polymer resin systems including polurethanes and polyvinyl chlorides, wherein the systems are already otherwise flame retardant, i.e. without regard to the bromobenzoate incorporated.

Mixtures of bromobenzoates containing a quantity of phthalate and smaller amounts of other impurities can be used in the invention. Mixtures containing 50% or more bromobenzoate, and preferably 65% or more bromobenzoate, are within the scope of this patent and are further described in the following examples.

Reference will now be made to specific examples using the processes described above. It is to be understood that the examples are provided to more completely describe preferred embodiments, and that no limitation to the scope of the invention is intended thereby.

Example 1Synthesis of 2-ethylhexyltetrabromobenzoate from
2,3,4,5-tetrabromobenzoic acid.

5 2,3,4,5-Tetrabromobenzoic acid (438g, 1.0 mole)
prepared via the decarboxylation of tetrabromophthalic
anhydride, 2-ethylhexanol (195g, 1.5 moles) and titanium
isopropoxide (2.2g, 0.008 mole) were charged to a stirred,
10 glass vessel. The mixture was heated at 200°C in nitrogen
atmosphere under stirring for 8 hours with the water of
reaction being collected in a Dean-Stark trap. The
resulting dark-brown reaction mixture was cooled to 90°C
and neutralized by stirring with solid Na₂CO₃ 10H₂O (5 wt%
15 on the reaction mixture) at 90°C for 1 hour. Following a
steam-strip to remove excess 2-ethylhexanol, the product
was filtered to remove the insoluble solid contaminants. A
clear, dark-brown liquid was obtained (533g, 97% yield).
This crude product was distilled at 192-194°C and 0.1-0.2
20 mmHg. A total of 527g of pale yellow, clear liquid was
obtained (99% recovery). GLC assay 96.2%
tetrabromobenzoate; OBr 56.6% (58.1% theory); acidity 0.15
meq.KOH/100g; TGA 5% wt. loss at 271°C; Gardner Color 3.

25

Example 2

5 Synthesis of 2-ethylhexyldibromobenzoate from
 dibromophthalic anhydride in 2-ethylhexanol solvent using
 sodium bicarbonate catalyst

Dibromophthalic anhydride (153g, 0.50 moles), 2-ethylhexanol (196 grams, 1.5 moles), and sodium bicarbonate (3.8g, 0.045 moles) are charged to a stirred, glass vessel. The mixture is heated to reflux, with the water of reaction being collected in a Dean-Stark trap. After CO₂ evolution ceases, the reaction is cooled and filtered to remove the catalyst. Excess 2-ethylhexanol is removed vis vacuum distillation to yield a liquid product. Product organic
15 bromine content is about 38%.

Example 3

20 Synthesis of 2-ethylhexyltetrabromobenzoate from
 tetrabromophthalic anhydride in 2-ethylhexanol solvent
 using sodium bicarbonate catalyst.

Tetrabromophthalic anhydride (232g, 0.50 moles), 2-ethylhexanol (196g, 1.5 moles), and sodium bicarbonate (3.8g, 0.045 moles) were charged to a stirred, glass vessel. The mixture was heated to reflux, with the water of reaction being collected in a Dean-Stark trap. After CO₂ evolution ceased (approx. 3.5 hours), the reaction was
30 cooled and filtered to remove the catalyst. The excess 2-ethylhexanol was stripped off under vacuum yielding a clear, amber liquid product. GLC assay indicated 76.1% tetrabromobenzoate and 13.0% tetrabromophthalic diester; organic bromide ("OBr") of 55.16%; TGA 5% wt. loss 217°C;
35 Gardner color 11.

Example 4

40 Synthesis of 2-ethylhexyltetrabromobenzoate from
 tetrabromophthalic anhydride in 2-ethylhexanol solvent
 using potassium bicarbonate catalyst.

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Tetrabromophthalic anhydride (232g, 0.50 moles), 2-ethylhexanol (196g, 1.5 moles), and potassium bicarbonate (5.5g, 0.055 moles) were charged to a stirred, glass vessel. The mixture was heated to reflux, with the water
5 of reaction being collected in a Dean-Stark trap. After CO₂ evolution ceased (approx. 3.5 hours), the reaction was cooled and filtered to remove the catalyst. The excess 2-ethylhexanol was stripped off under vacuum yielding a clear, amber liquid product. GLC assay 66.7%
10 tetrabromobenzoate, 15.8% tetrabromophthalic diester; OBr 54.04%; TGA 5% wt. loss 221°C; Gardner color 12.

Example 5

Synthesis of 2-ethylhexyltetrabromobenzoate from
tetrabromophthalic anhydride in 2-ethylhexanol solvent
using potassium carbonate catalyst.

5
Tetrabromophthalic anhydride (464g, 1.0 moles), 2-ethylhexanol (391g, 3.0 moles), and potassium carbonate (15.1g, 0.11 moles) were charged to a stirred, glass
10 vessel. The mixture was heated to reflux, with the water of reaction being collected in a Dean-Stark trap. After CO₂ evolution ceased (approx. 3.5 hours), the reaction was cooled and filtered to remove the catalyst. The excess 2-ethylhexanol was stripped off under vacuum yielding a
15 clear, amber liquid product. GLC assay 63.5% tetrabromobenzoate, 11.5% tetrabromophthalic diester; OBr 52.29%; THA 5% wt. loss 235°C; Gardner color 14.

Example 6

Synthesis of 2-ethylhexyltetrabromobenzoate from
tetrabromophthalic anhydride in 2-ethylhexanol solvent
using lithium carbonate catalyst.

25
Tetrabromophthalic anhydride (464g, 1.0 moles), 2-ethylhexanol (390g, 3.0 moles), and lithium carbonate (8.1g, 0.11 moles) were charged to a stirred, glass vessel. The mixture was heated to reflux, with the water of
30 reaction being collected in a Dean-Stark trap. After CO₂ evolution ceased (approximately 8.0 hours), the reaction was cooled and filtered to remove the catalyst. The excess 2-ethylhexanol was stripped off under vacuum yielding a
clear, amber liquid product. GLC assay 60.7%
35 tetrabromobenzoate, 10.4% tetrabromophthalic diester; OBr 51.9%; TGA 5% wt. loss 240°C.

Example 7

40
Synthesis of 2-ethylhexyltetrabromobenzoate from
tetrabromophthalic anhydride in 2-ethoxyethylether solvent.

Tetrabromophthalic anhydride (1391g., 3.00 moles), 2-ethylhexanol (469g, 3.60 moles), 2-ethoxyethylether (771g, 4.75 moles), and sodium bicarbonate (25g, 0.30 moles) were
5 charged to a stirred, glass vessel. The mixture was brought to reflux, with the water of reaction being collected in a Dean-Stark trap. After CO₂ evolution had ceased (approximately 3 hours), the reaction was cooled and filtered to remove the catalyst. The 2-ethoxyethylether
10 and excess 2-ethylhexanol were stripped off under vacuum yielding a clear, amber liquid product. GLC assay 85.0% tetrabromobenzoate, 1.6% tetrabromophthalic diester; OBr 56.99%; TGA 5% wt. loss 209°C, Gardner color 11.

Example 8Synthesis of 2-butoxyethyltetrabromobenzoate.

5 Tetrabromophthalic anhydride (232g, 0.5 moles), 2-butoxyethanol (226g, 1.9 moles), and potassium bicarbonate (8.3g, 0.083 moles) were charged to a stirred, glass vessel. The mixture was heated to reflux, with the water of reaction being collected in a Dean-Stark trap. After
10 CO₂ evolution had ceased (approximately 3 hours), the reaction was cooled and filtered to remove the catalyst. Excess 2-butoxyethanol was stripped off under vacuum yielding a clear, amber product. GLC assay 82.4% tetrabromobenzoate, 7.9% tetrabromophthalic diester; OBr
15 57.84%; TGA 5% wt. loss 225°C; Gardner color 13.

Example 9Synthesis of 2-methoxyethoxyethyltetrabromobenzoate.

20 Tetrabromophthalic anhydride (232g, 0.5 moles), 2-methoxyethoxyethanol (238g, 2.0 moles), and potassium bicarbonate (8.3g, 0.083 moles) were charged to a stirred, glass vessel. The mixture was heated to reflux, with the
25 water of reaction being collected in a Dean-Stark trap. After CO₂ evolution had ceased (approximately 2 hours), the reaction was cooled and filtered to remove the catalyst. Excess 2-methoxyethoxyethanol was stripped off under vacuum yielding a clear, amber product. GLC assay 81.2%
30 tetrabromobenzoate, 0.4% tetrabromophthalic diester; OBr 59.82%; TGA 5% wt. loss 180°C; Gardner color 14.

Example 10Synthesis of tetrabromobenzoates from mixed C₂-C₂ alcohols.

35 Tetrabromophthalic anhydride (464g, 1.0 moles), BASF C-C₂ alcohol (500 mls), and sodium bicarbonate (8.4g, 0.10 moles) were charged to a stirred, glass vessel. The
40 mixture was heated to reflux, with the water of reaction being collected in Dean-Stark trap. After CO₂ evolution had ceased (approximately 3.5 hours), the reaction was

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cooled and filtered to remove the catalyst. Excess alcohol was stripped off under vacuum yielding a clear, golden colored product. OBr 52.77%; TGA 5% wt. loss 231°C; Gardner color 5.

5

Example 11Use of a tetrabromobenzoate in flexible polyurethane foam.

10

The 2-ethylhexyltetrabromobenzoate of Example 3 was added to flexible polyurethane foam prepared by the formulation below. All components are expressed in parts by weight.

15

<u>Component</u>	<u>Parts by Weight</u>
3,000 molecular weight glycerine based heteropolyol	100
2-ethylhexyltetrabromobenzoate of Example 3	18
Water	4.5
Amine catalyst (triethylene diamine/dimethylaminoethyl ether)	0.33
L-620 silicone surfactant*	0.9
Stannous octoate	0.25
Toluene diisocyanate (80/20)	51.2

35 *product of OSI Specialties

Lab preparation yielded a flexible polyurethane foam with a density of 1.5 lb/ft² and an airflow of 4.0. The foam met the criteria of the California Bulletin 117 combustibility test part A with an average burn distance of 2.5 inches and no burn time. It also met the criteria of

40

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part D with a weight retention in the smolder test in excess of 98%.

5

Example 12Use of 2-Ethylhexyltetrabromobenzoate/aryl phosphate blend in flexible polyether polyurethane foam.

10 The bromobenzoate of Example 3 was added to a flexible polyether polyurethane foam similar to that of the previous example except that the water content was increased to lower the density. A lower density foam has the characteristic of being more difficult to flame retard. To
15 further enhance the flammability test performance, 6.7 parts of a triaryl phosphate were added. This phosphate can be added separately or blended with the bromobenzoate to give an enhanced product.

20	<u>Component</u>	<u>Parts by Weight</u>
	3,000 molecular weight glycerine based heteropolyol	100
25	2-ethylhexyltetrabromobenzoate of Example 3	20
	Water	6.6
30	Amine catalyst (triethylene diamine/dimethylaminoethyl ether)	0.6
	L-620 silicone surfactant*	0.8
35	Stannous octoate	0.35
	Toluene diisocyanate (80/20)	79.46

*product of OSI Specialties

40

The resulting foam had a density of 1.3 lb/cu. ft. and an air flow of 6.7 Scfm. The foam met the criteria of California Bulletin 117 with an average void length of 0.5

-17-

inches, 0 seconds burn time in the open flame test. The foam also met the criteria of MVSS-302, a flammability test for components of automotive interiors.

A foam prepared which contained the aryl phosphate without the bromobenzoate of Example 3 failed the California Bulletin 117 flammability test. The burn time exceeded 25 seconds and the void length was greater than 10 inches. The density of this foam was 1.3 lb/cu. ft. with an air flow of 5.1 Scfm.

10

Example 13

Use of other tetrabromobenzoates in flexible polyether polyurethane foam.

15

The 2-butoxyethyltetrabromobenzoate of Example 8 and the 2-methoxyethoxyethyltetrabromobenzoate of Example 9 were also observed to perform well as flame retardants in flexible polyether polyurethane foam. The formulation below differed slightly from those of Examples 11 and 12 above.

25	<u>Component</u>	<u>Parts by Weight</u>
	3,000 molecular weight glycerine based heteropolyol	100
30	Tetrabromobenzoate of Example 8 or 9	23
	Water	4.5
	Amine catalyst (triethylene diamine/dimethylaminoethyl ether)	0.5
35		
	L-620 silicone surfactant*	0.9
	Stannous octoate	0.4
40	Toluene diisocyanate (80/20)	57.15

*product of OSI Specialties

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Foams having a density of 1.35 lb/cu. ft. were produced. The air flow of the foam containing the bromobenzoate of Example 8 was 4.9 Scfm. The foam
5 containing the tetrabromobenzoate of Example 9 was more open celled (and hence more flammable) with an air flow of 6.3 Scfm.

The foam containing the tetrabromobenzoate of Example 8 met the criteria of California Bulletin 117 with an average
10 void length of 3.35 inches. The foam containing the tetrabromobenzoate of Example 9 had one test specimen of five which burned in excess of 8 inches.

15 Example 14

Use of tetrabromobenzoate from mixed C₇-C₉ alcohols in flexible polyether polyurethane foam.

20 The tetrabromobenzoate of Example 10 was also incorporated into flexible polyether polyurethane foam. It was added to a low density foam in the form of a blend with aromatic phosphate ester to enhance performance.

25

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	<u>Component</u>	<u>Parts by Weight</u>
5	3,000 molecular weight glycerine based heteropolyol	100
	75/25 blend benzoate of Example 10/ triaryl phosphate ester	27
	Water	7.6
10	Amine catalyst (triethylene diamine/dimethylaminoethyl ether)	0.5
	L-620 silicone surfactant*	1.0
15	Stannous octoate	0.36
	Toluene diisocyanate (80/20)	90.11
20	<hr/>	
	*product of OSI Specialties	

To generate a density of 1.08 lb/cu. ft., 5 parts of methylene chloride was added as an auxiliary blowing agent. The air flow was 8.6 Scfm. This very flammable foam met California Bulletin 117 criteria with 27 parts of the benzoate blend. The average void length was 6.3 inches with 0 seconds burn time.

30 Example 15

Use of tetrabromobenzoate in flexible polyester flexible polyurethane foam.

35 The tetrabromobenzoate of Example 3 was added to flexible polyester polyurethane foam prepared using the formulation below. Its performance was compared to that of Tris (dichloropropyl) phosphate, used at the same loading level.

40	<u>Component</u>	<u>Parts by Weight</u>
	Witco Fomrez 53 polyester polyol	100

-20-

	Bromobenzoate of Example 3 or Tris (dichloropropyl) phosphate	8
5	Water	4.0
	Amine catalyst 1	0.6
	L-532 silicone surfactant*	1.3
10	Amine catalyst 2	0.2
	Toluene diisocyanate (80/20)	50.4
15	<hr/> *product of OSI Specialties	

Both formulations produced foams which had densities of 2.1 lb/cu. ft. Both met MVSS-302 flammability criteria with an SE (highest) rating. The foam containing the bromobenzoate had better tensile strength initially 19.4 lb/sq. in. vs. 16.5 lb/sq. in. for the foam containing Tris (dichloropropyl) phosphate. It also maintained its tensile properties better after 24 hours humid aging (70°C), indicating better hydrolytic stability for the benzoate.

25

Example 16Use of tetrabromobenzoate in reaction injection molded polyurethane.

5

A commercial flame retardant reaction injection molded (RIM) polyurethane system was used for evaluation of the 2-ethylhexyltetrabromobenzoate of Example 3. The standard RIM system contains 15% of a blend of pentabromodiphenyl oxide and an aromatic phosphate ester (DE-60FS) as a combustibility modifier. RIM parts were molded containing this standard combustion modifier and containing the same level (15%) of a 2:1 blend of 2-ethylhexyltetrabromobenzoate and triethyl phosphate (BB/TEP). Results are tabulated below.

15

Combustion Modifier	UL-94 Rating	Heat Distortion Temperature
15% DE60FS	V-0	59°C
15% BB/TEP	V-0	62°C

20

25

Both combustion modifiers allowed the polymer system to meet the desired combustibility standard, UL-94 V-0, but the blend containing the tetrabromobenzoate maintained the key physical property, heat distortion, better than the standard combustibility modifier.

30

Example 17Use of 2-Ethylhexyltetrabromobenzoate in rigid
polyurethane foam.

5 The 2-ethylhexyltetrabromobenzoate of Example 3 was
added to a rigid polyurethane foam as indicated in the
formulation below. The resultant foam, which had good cell
structure and a density of 2.1 lbs. per cubic foot, was
10 tested for combustibility in a four foot tunnel designed to
correlate with the results obtained on a larger scale in
the 25 foot tunnel test (ASTM E-84). The four foot tunnel
test consists of a Fisher burner with a 1950°F butane flame
placed under a 4 foot long specimen inclined at an angle of
15 fifteen degrees. Smoke is measured by passing combustion
products between a light beam and photocell. Standards
with known 25 foot tunnel test values are used for
calibration.

20	<u>Component</u>	<u>Parts by Weight</u>
	Sucrose polyol (N500 OH#)	34
	Tetrabromophthalate Diol (N215 OH#)	51
25	2-ethylhexyltetrabromobenzoate of Example 3	15
	L-5440 surfactant (product of OSI Spec)	1.5
30	Dimethylcyclohexylamine	0.8
	Dichlorofluoroethane	24.7
35	Polymeric MDI (N125 index)	80.6

Combustibility results of four foot tunnel testing of
this foam were a predicted flame spread of 19 and smoke
evolution of 170. Both are well within the requirements
40 for Class 1 of less than 25 flame spread and 450 smoke for
the 25 foot tunnel. Although tetrabromophthalate diol
contributes to the combustibility test performance, over 60
parts of this diol are needed in this formulation when the

-23-

tetrabromobenzoate is not present to attain less than 25 flame spread in the four foot tunnel test.

Example 18

5

Use of 2-Ethylhexyltetrabromobenzoate in unsaturated polyester resin.

10 To an orthophthalic/propylene glycol based resin containing 35% styrene was added 14.5 parts per hundred resin (phr) 2-ethylhexyltetrabromophthalate of Example 3 and 2 phr antimony oxide. An eighth inch thick, 25% glass laminate was prepared using 1.5 oz/yd² glass mat. This laminate had a V-0 rating when tested by the UL-94
15 combustibility test. In contrast, the resin without bromobenzoate fails the UL-94 combustibility test.

Example 19Use of 2-Ethylhexyltetrabromobenzoate in PVC.

5 Bromobenzoates are useful in polyvinyl chloride (PVC) both as flame retardants and as plasticizers. Many applications require both flexibility and enhanced flame resistance. The bromobenzoates of Examples 3 and 10 were compared to brominated dioctylphthalate as a secondary
 10 plasticizer in the formulation below at 20 parts by weight. Processing was on a two roll mill at 420°F for two minutes. Samples were compression molded at 350°F.

	<u>Component</u>	<u>Parts by Weight</u>
15	Geon Corp. Geon 30 PVC resin	100
	Lead stabilizer	5
20	Lubricant	0.2
	Antimony oxide	2.0
	Clay	30
25	Alumina trihydrate	20
	Phthalate plasticizer	30

30 Plasticizing characteristics were compared by measuring stress, strain, and modulus of tensile bars. Flammability was assayed using the Oxygen Index Test.

35		Avg. Stress @ Max Load	%Strain @ Auto Break	Modulus (psi)
	Bromobenzoate of Example 3	2400	250	5000
40	Brominated DOP	2200	350	3600

-25-

Bromobenzoate of Example 10	2400	270	5200
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The presence of bromobenzoate increased the modulus while reducing the strain. However, there was little difference in the stress at break. The oxygen index improved when the bromobenzoates were included.

Example 20

Use of tetrabromobenzoate in epoxy resin.

An epoxy resin consisting of an epichlorohydrin-bisphenol A condensation product is blended with the bromobenzoate of Example 8 and a quantity of antimony oxide synergist. To this blend is slowly added a quantity of amine hardner over several minutes. The mixture is then poured into 1/16" thick molds and placed into a 55°C oven overnight (16 hours).

A further cure cycle of 2 hours at 125°C followed by two hours at 155°C is followed by cooling to room temperature and cutting test specimens to size. Two flammability tests are performed, oxygen index and UL-94. Results appear below.

	<u>FLAME RETARDANT</u>	<u>OXYGEN INDEX</u>	<u>UL-94 RATING</u>
	None	27	Fail
30	10% bromobenzoate	33	----
	13.3% bromobenzoate	35	----
	16% bromobenzoate/ 4% antimony oxide	---	V-O

It can be seen from the above that bromobenzoates are effective flame retardants in epoxy resin systems.

Example 21

40

Use of a tetrabromobenzoate in rubber.

The tetrabromobenzoate of Example 9 (from mixed C₇-C₉ alcohols) is added to a rubber conveyor belt cover. The rubber is a 50/50 blend of cis-polyisoprene and cis-polybutadiene with 70 parts N-220 carbon black and 40 parts naphthenic oil. A second rubber is formulated substituting 30 parts of the tetrabromobenzoate of Example 9 for 30 parts of the oil.

Bars measuring 1/2" x 5" x 1/8" are tested by limiting oxygen index (ASTM D-2863). The rubber containing 40 parts oil has an oxygen index of 20. Substituting 30 parts of the brominated benzoate of Example 9 raises the oxygen index to 23.

Tensile strength increases only slightly, from 3300 to 3500 psi with the substitution of the benzoate of Example 9. Shore A hardness goes from 65 to 70.

While the invention has been illustrated and described in detail in the foregoing description, the same is to be considered as illustrative and not restrictive in character, it being understood that only the preferred embodiment has been described and that all changes and modifications that come within the spirit of the invention are desired to be protected.

CLAIMS

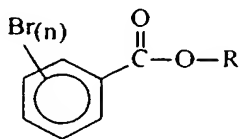
We claim:

5

1. A method of providing flame retarded polyurethanes, comprising including a bromobenzoate compound as a component of the polyurethane mixture.

10

2. A method according to claim 1 wherein said bromobenzoate is of the formula:



15

wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

20

3. A method according to claim 1 wherein said polyurethane is a flexible polyurethane.

4. A method according to claim 1 wherein said polyurethane is a flexible polyether polyurethane.

25

5. A method according to claim 1 wherein said polyurethane is a flexible polyester polyurethane.

30

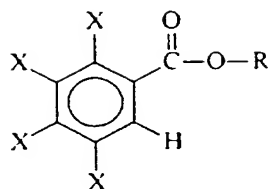
6. A method according to claim 1 wherein said polyurethane is a reaction injection molded (RIM) polyurethane system.

7. A method according to claim 1 wherein said polyurethane is a rigid polyurethane.

35

8. A method according to claim 2 wherein said tetrabromobenzoate compound is of the formula:

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wherein each X is -Br or -H, with the proviso that at least one X is -Br.

5

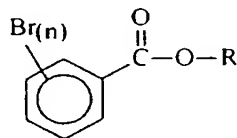
9. The method of claim 8, wherein each X is -Br.

10. The method of claim 9, wherein said compound is selected from the group 2-ethylhexyltetrabromo-benzoate, 2-butoxyethyltetrabromo-benzoate, and 2-methoxyethoxyethyl-tetrabromobenzoate.

11. A method of improving the flame retardency of unsaturated polyester resins comprising incorporating a bromobenzoate compound into the unsaturated polyester resin.

12. A method according to claim 11 wherein said bromobenzoate compound is of the formula:

20

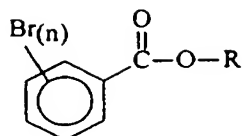


wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

25

13. A method of providing flame retarded polyvinyl chloride, comprising incorporating a bromobenzoate compound into the polyvinyl chloride.

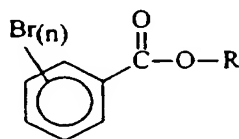
14. A method according to claim 13 wherein said bromobenzoate is of the formula:



wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

- 5 15. A method of providing a plasticizer to a polyvinyl chloride, comprising incorporating a bromobenzoate compound into the polyvinyl chloride.

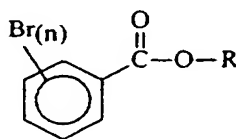
16. A method according to claim 15 wherein said
10 bromobenzoate compound is of the formula:



wherein n is an integer from 1 to 5 and R is an organic
15 group having up to about 30 carbon atoms.

17. A method of improving the flame retardency of epoxy resins comprising incorporating a bromobenzoate compound into the epoxy resin.

- 20 18. A method according to claim 17 wherein said bromobenzoate compound is of the formula:



25 wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

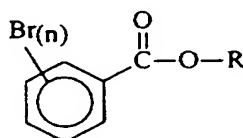
19. A method according to claim 17 wherein said
30 bromobenzoate compound is a tetrabromobenzoate compound.

20. A method according to claim 19 wherein said tetrabromobenzoate compound is 2-methoxyethoxyethyltetra-bromobenzoate.

21. A method of improving the plasticity of rubber, comprising incorporating a bromobenzoate compound into the rubber.

5

22. A method according to claim 21 wherein said bromobenzoate is of the formula:



10

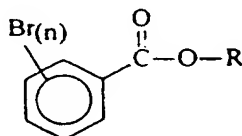
wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

23. A flame retarded polyurethane composition, comprising a polyurethane polymer and an effective flame retarding amount of a bromobenzoate compound.

15

24. A composition according to claim 23 wherein said bromobenzoate compound is of the formula:

20



wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

25

25. A composition according to claim 23 wherein said polyurethane is a reaction injection molded polyurethane.

26. A composition according to claim 24 wherein said bromobenzoate compound is 2-ethylhexyltetrabromobenzoate.

30

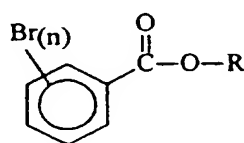
27. A composition according to claim 24 wherein said bromobenzoate compound is 2-butoxyethyltetrabromobenzoate.

35

28. A composition according to claim 24 wherein said bromobenzoate compound is 2-methoxyethoxyethyltetra-bromobenzoate.

5 29. A flame retarded polyester composition, comprising an unsaturated polyester resin incorporating an effective flame retarding amount of a bromobenzoate compound.

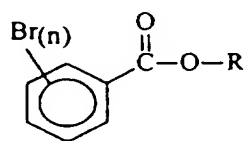
30. A composition according to claim 29 wherein said
10 bromobenzoate compound is of the formula:



wherein n is an integer from 1 to 5 and R is an organic
15 group having up to about 30 carbon atoms.

31. A flame retarded polyvinylchloride composition, comprising a polyvinylchloride polymer incorporating an effective flame retarding amount of a bromobenzoate
20 compound.

32. A composition according to claim 31 wherein said bromobenzoate compound is of the formula:

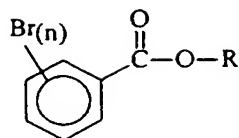


25

wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

33. A plasticized polymer composition, comprising polyvinyl chloride incorporating an effective plasticizing amount of a bromobenzoate compound.

34. A composition according to claim 33 wherein said
35 bromobenzoate compound is of the formula:

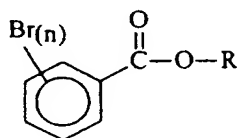


wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

5

35. A flame retardand epoxy composition, comprising an epoxy resin incorporating an effective flame retardant amount of a bromobenzoate compound.

10 36. A composition according to claim 35 wherein said bromobenzoate compound is of the formula:



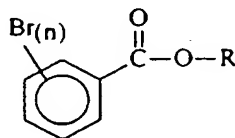
15 wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

37. A composition according to claim 36 wherein said bromobenzoate compound is 2-methoxyethoxyethyltetra-
20 bromobenzoate.

38. A plasticized rubber composition, comprising rubber and an effective plasticizing amount of a bromobenzoate compound incorporated into the rubber.

25

39. A composition according to claim 38 wherein said bromobenzoate compound is of the formula:



30

wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

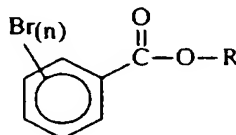
40. A flame retarded polymer composition, comprising:

a polymer resin selected from the group consisting of polyurethane, polyvinyl chloride, and epoxy resins; and

a flame retardant system including a brominated organic flame retardant component including one or more brominated organic flame retardants, wherein said brominated organic flame retardant component is at least about 50% by weight comprised of one or more bromobenzoate compounds.

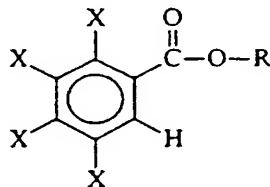
41. The composition of claim 40 wherein said brominated organic flame retardant component consists essentially of one or more bromobenzoate compounds.

42. The composition of claim 41 wherein said one or more bromobenzoate compounds are encompassed by the formula:



wherein n is an integer from 1 to 5 and R is an organic group having up to about 30 carbon atoms.

43. A composition of claim 42 wherein said one or more bromobenzoate compounds are encompassed by the formula:



wherein each X is $-\text{Br}$ or $-\text{H}$, with the proviso that at least one X is $-\text{Br}$.

44. A composition according to claim 43 wherein each X is $-\text{Br}$.

45. A composition according to claim 44 wherein said bromobenzoate compound is selected from the group

consisting of 2-ethylhexyltetrabromobenzoate, 2-
butoxyethyltetrabromo-benzoate, and 2-
methoxyethoxyethyltetrabromobenzoate.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US96/04976

A. CLASSIFICATION OF SUBJECT MATTER

IPC(6) : C08J 5/10; C08K 5/09, 5/10, 5/12; C08L 75/04

US CL : 524/288, 290, 291, 292

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 524/288, 290, 291, 292

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

NONE

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US, A, 5,049,697 (BOHEN et al.) 17 September 1991, entire document.	1-45
Y	US, A, 5,329,054 (THERIOT et al.) 12 July 1994, entire document.	1-45

☐ Further documents are listed in the continuation of Box C. ☐ See patent family annex.

* Special categories of cited documents:	*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
A Document defining the general state of the art which is not considered to be part of particular relevance	*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
E earlier document published on or after the international filing date	*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
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P Document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

12 JULY 1996

Date of mailing of the international search report

27 AUG 1996

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